

**When does a crystal conduct heat like a glass?**

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Motivated by the search for improved thermoelectric materials, several compounds have attracted attention that combine the high electron mobilities found in crystals with a low thermal conductivity  $\kappa$ , approaching  $\kappa$ -values typical for glasses. The common structural feature of these “electron crystal phonon glass” (ECPG) materials is that they contain loosely bound atoms that reside in a large crystalline “cage”. These “rattlers” scatter phonons and greatly reduce the thermal conductivity of the material, but generally not to quite glasslike levels. A particular class of clathrates is formed by the filled skutterudite antimonides [1]. More recently, the Ge-clathrate  $\text{Sr}_8\text{Ga}_{16}\text{Ge}_{30}$  was found to be an ECPG material, having a truly glasslike thermal conductivity while maintaining the crystalline electronic properties [2, 3].

Here we report a combination of Resonant Ultrasound Spectroscopy, neutron-scattering, low-temperature ultrasonic attenuation and thermal conductivity measurements for filled and unfilled skutterudites and for Ge-clathrates. Taken together, these measurements suggest specific structural features that result in a crystal with the lowest possible thermal conductivity, namely that of a glass with the same chemical composition. The weakly bound atoms that “rattle” within the oversized atomic cages in the crystal result in a low thermal conductivity, but the present data show that both “rattling” *and* tunneling states are necessary to produce a true glass-like thermal conductivity.

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[1] V. Keppens et al., *Nature* 395, 876-878 (1998).

[2] J.L. Cohn et al., *Phys. Rev. Lett.* 82, 778-782 (1999).

[3] V. Keppens et al., *Phil. Mag. Lett.* 80, 807-812 (2000).

